

Geochemistry of volcanic and hydrothermal gases of Mutnovsky volcano, Kamchatka: evidence for mantle, slab and atmosphere contributions to fluids of a typical arc volcano

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Abstract We report chemical compositions (major and trace components including light hydrocarbons), hydrogen, oxygen, helium and nitrogen isotope ratios of volcanic and geothermal fluids of Mutnovsky volcano, Kamchatka. Several aspects of the geochemistry of fluids are discussed: chemical equilibria, mixing of fluids from different sources, evaluation of the parent magmatic gas composition and contributions to magmatic vapors of fluids from different reservoirs of the Kamchatkan subduction zone. Among reactive species, hydrogen and carbon monoxide in volcanic vapors are chemically equilibrated at temperatures $>300^{\circ}\text{C}$ with the $\text{SO}_2\text{-H}_2\text{S}$ redox-pair. A metastable equilibrium between saturated and unsaturated light hydrocarbons is attained at close to discharge temperatures. Methane is disequilibrated. Three different sources of fluids from three fumarolic fields in the Mutnovsky craters can be distinguished: (1) magmatic gas from a large convecting

magma body discharging through Active Funnel, a young crater with the hottest fumaroles (up to 620°C) contributing $\sim 80\%$ to the total volcanic gas output; (2) volcanic fluid from a separate shallow magma body beneath the Bottom Field of the main crater ($96\text{--}280^{\circ}\text{C}$ fumaroles); and (3) hydrothermal fluid with a high relative and absolute concentrations of CH_4 from the Upper Field in the main crater ($96\text{--}285^{\circ}\text{C}$ fumaroles). The composition of the parent magmatic gas is estimated using water isotopes and correlations between He and other components in the Active Funnel gases. The He-Ar- N_2 systematics of volcanic and hydrothermal fluids of Mutnovsky are consistent with a large slab-derived sedimentary nitrogen input for the nitrogen inventory, and we calculate that only $\sim 1\%$ of the magmatic N_2 has a mantle origin and $\ll 1\%$ is derived from the arc crust.

Keywords Volcanic gases · He and N isotopes · Subduction · Recycling · Hydrothermal systems

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Introduction

Degassing of volatiles through active arc volcanoes is a direct manifestation and a measureable property of the volatile recycling in subduction zones. The fluxes of sedimentary material and fluids from the subducting slab are responsible for the genesis of arc magmas and the occurrence of volcanism along convergent plate boundaries. The estimates of production rates of major and trace gas components, using instrumentally measured SO_2 fluxes, have shown that the amount of volatiles released from volcanoes are to a large extent balanced by those supplied from various sources, such as the crust, the mantle and subducted material (Andres and Kasgnoc 1998; Hilton et al. 2002; Wallace 2005; Mather et al. 2006; Taran 2009). A

number of workers have used different methods for distinguishing potential contributions to the total volatile input/output—the mantle wedge, arc crust or the sedimentary and basaltic layers of the subducting slab (Varekamp et al. 1992; Sano and Marty 1995; Fischer et al. 1998, 2002, 2008; Sano et al. 1998, 2001; Hilton et al. 2002; Shaw et al. 2003; Wallace 2005; Taran 2009). Fischer et al. (1998) used chemical and isotopic composition of CO₂, N₂ and He in gases of Kudryavy volcano, located at the southern part of the Kurilian arc, to show that for this part of the Kamchatka-Kurile subduction zone, the volatiles discharged from this single volcano could be supplied from the subducted slab and mantle wedge alone, without contribution from the underlying arc crust.

Mutnovsky volcano is another typical subduction-type volcano at the Kamchatka-Kurile subduction zone, located about 1000 km to the north of Kudryavy, with continuous high-temperature fumarolic activity and frequent phreatic eruptions. The volcano is characterized by a series of thermal manifestations in its craters including high-temperature (>600°C) fumaroles, ultra-acidic boiling pools, a crater lake, steaming grounds and spectacular 600-m high crater walls of altered rocks covered by a thick glacier (Fig. 1). On its northern slopes several groups of steam vents and thermal springs demonstrate the presence of a powerful geothermal system. Almost all the fumarolic fields of the volcano are easily accessible except high-temperature fumaroles from so-called “Active Funnel”, the youngest, 60-m deep, 250-m wide crater with nearly vertical walls and a permanent ~1 km-high gas plume.

Mutnovsky is one of the best studied volcanoes in Kamchatka. Soon after the 1960–61 phreatic eruption, two papers on the gas composition were published (Vakin et al. 1966; Serafimova 1966). Serafimova (1966) presented a detailed map of the Bottom fumarolic field of the main crater showing temperatures and flow rates of fumaroles, and reported 9 chemical analyses of the hottest fumaroles, with two analyses from the Active Funnel. Polyak et al. (1985) and Muraviev et al. (1983) have estimated the heat output from the volcano. Taran et al. (1986, 1987) studied chemistry and isotopic compositions of hydrothermal fluids from natural manifestations and the first geothermal wells of the Mutnovsky geothermal field. Taran et al. (1992) published a study of the Mutnovsky fumarolic system with a conceptual geochemical model of the volcano-hydrothermal activity of the volcano. They presented chemical and isotopic data for one more fumarole from the Active Funnel, with a temperature of 360°C.

Here we report our data on the gas geochemistry obtained as the result of extensive sampling of all fumarolic fields in the volcano craters and main geothermal manifestations and wells of the geothermal field performed during the 2001 and 2006–2007 field seasons. More than 20

samples were taken from fumaroles of Active Funnel. For the first time a comprehensive data set is presented on light hydrocarbons in volcanic and hydrothermal gases of Mutnovsky, as well as the He and Ar concentrations and nitrogen isotopic composition.

One of the main goals of this study is to estimate the sources of volatiles contributing to the Mutnovsky volcanic and hydrothermal fluids using a large set of geochemical data that includes the composition of major (H₂O, CO₂, S, HCl, N₂, H₂) and trace (Ar, He, CO, CH₄, and C₂–C₄ hydrocarbons) species, ³He/⁴He, D/H, ¹⁸O/¹⁶O and ¹⁵N/¹⁴N isotopic ratios. We attempt to derive the composition of the parent magmatic gas and to estimate the mantle, crust and slab fractions in the magmatic discharge of Mutnovsky.

General setting

Mutnovsky volcano is located ~75 km to the south of the main city of the Kamchatka district Petropavlovsk-Kamchatsky (Fig. 1). It has an elongated edifice composed of merged stratocones (Fig. 2a). The highest point is 2323 m asl (52.35 N, 158.27 E). Two large joined craters, Northern and Southwest, of 1.3×1.3 and 1.5×2 km², respectively, are situated ~1 km westward from the main summit. The young active crater with the most intensive emissions, named “Active Funnel” cuts the rim of the large Southwest crater (Fig. 2b).

Mutnovsky is located ~20 km from the Pacific coast, ~200 km from the Kamchatkan trench and belongs to the Eastern (frontal) volcanic belt of Kamchatka (Fig. 1). The depth to the Mohorovicic discontinuity beneath Mutnovsky is 22 km, the depth to the Benioff zone is 90–110 km (Gorbatov et al. 1997). The first detailed geological description of Mutnovsky was reported by Marenina (1956). The volcano resides on a basement that is interpreted as a complex of Paleogene-Neogene volcanogenic-sedimentary rocks with domination of volcanogenic component ranging from basalts to rhyolites (Selyangin 1993). All these rocks are partly metamorphosed to greenschist facies and altered; propylitization is the most common type of the hydrothermal alteration. A large (15×15 km²) Miocene Akhomten granodiorite intrusion borders Mutnovsky from the east.

The Mutnovsky Geothermal Field extended 6–7 km to the north of the Mutnovsky volcano (Fig. 2a). Tens of geothermal wells, 1–2.5 km-deep, produce fluid from 240–300°C geothermal aquifer for operating a 60 MW power-plant.

According to Melekestsev et al. (1987), the last eruption of the volcano that provided juvenile basaltic material, occurred in the 1848. During the 20th century, six small phreatic

Fig. 1 Upper Field fumaroles in the crater of Mutnovsky volcano. Inset: location of Mutnovsky volcano, southern Kamchatka (red triangle). Triangles are Holocene and active volcanoes. PK Petropavlovsk-Kamchatsky

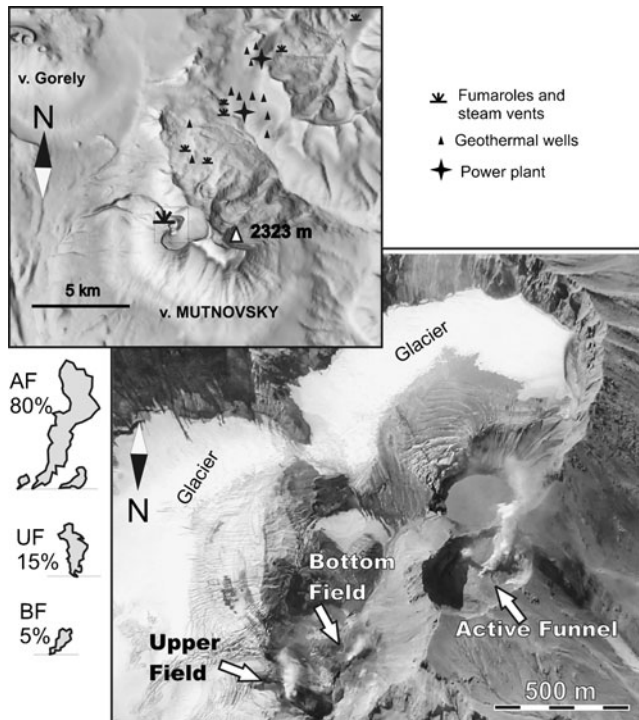
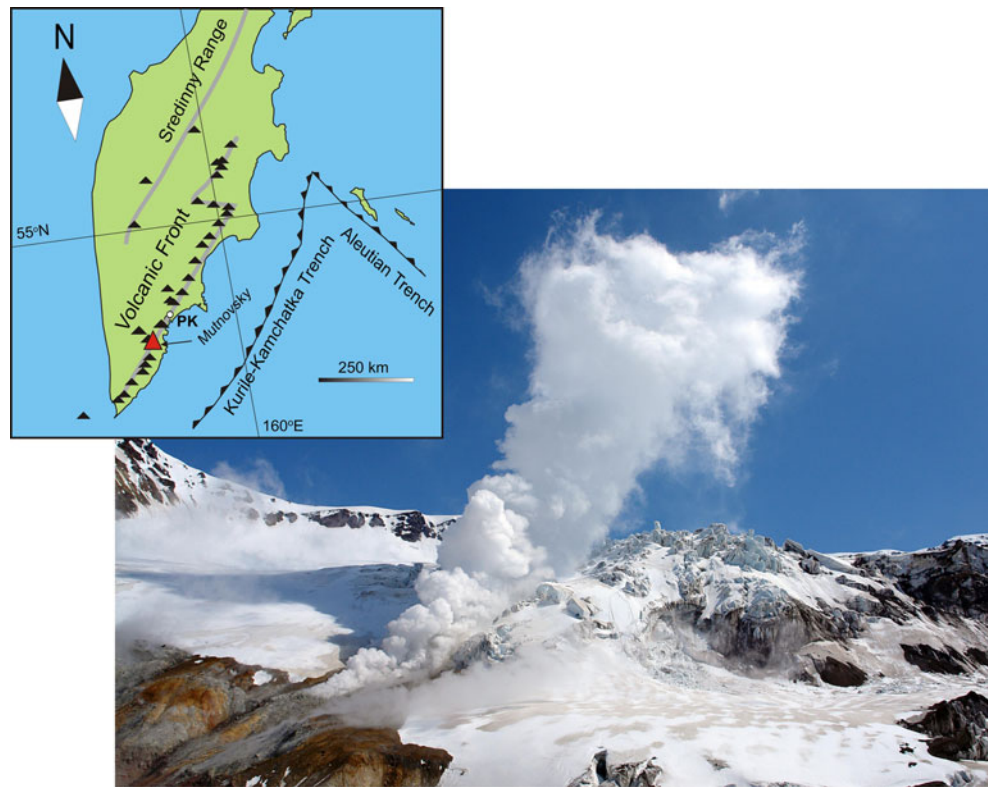


Fig. 2 a: Digital map of the Mutnovsky area with volcanic edifices and geothermal field. b: Craters of Mutnovsky volcano with fumarolic fields. The visible area of the contours of fumarolic plumes (left side of the down cartoon) may be used for estimation of the relative steam output (see text for explanations). AF, BF and UF are Active Funnel, Bottom Field and Upper Field, respectively

eruptions have been recorded: 1904, 1916–17, 1927–29, 1939–40, 1960–61 and 2000. The 1960–1961 and 2000 eruptions are well documented (Kirsanov 1964; Serafimova 1966; Vakin et al. 1966; Zelenski et al. 2002). The last phreatic explosion occurred on the 17th of April 2007.

Eruptive products of Mutnovsky have evolved from basalts to andesites and dacites; the most felsic rocks of the whole volcano are two rhyodacite flows (Selyangin 1993). Duggen et al. (2007) proposed the dehydration of the slab sediments and altered basalts as the main factor providing the parent magma generation of the Mutnovsky volcano. The slab component beneath Mutnovsky is inferred to be primarily a hydrous fluid with a minor amount of a silicate melt.

Sampling and analysis

Samples of fluids from five sampling sites were taken (Fig. 2). Two of them are located in the Northern Mutnovsky crater: the Bottom Field (BF) and the Upper Field (UF). The third site is Active Funnel (AF) which is the most recent crater of Mutnovsky with oval shape (~250 m in diameter) and steep to vertical walls from 60 to 200 m high. Fumaroles in this young crater are located on the bottom and within the 1/3 lowest part of the southern wall. The hottest vent had 620°C in 2006. Discharge conditions at the Bottom Field are quite various and include

boiling-point and hot fumaroles (up to 280°C), boiling springs, drainless boiling pools with some of them filled with the ultra-acidic brine; mud pools and bubbling gas in the Vulcannaya stream which drains all water discharges of the Northern crater. Upper Field is located on the western slope of the Northern crater, 20–50 m above the crater floor, close to the front of the glacier. There are several powerful “roaring” fumaroles there with temperatures up to 280°C (Fig. 1).

Volcanic and hydrothermal gases were sampled into 100–250 ml Pyrex ampoules with a Teflon stopcock filled with 50 ml of 4 N KOH solution, according to the procedure developed by Giggenbach (1975). At a sampling site, a 20×1000 mm silica tube was inserted into a fumarole vent. An inner 8 mm silica tube connected to short silicone rubber tubing was put into the outer tube to let gas flow freely with a flow rate enough to prevent any condensation. Additionally, 50 ml of gas condensate was collected at each sampling site by pumping gas through two sequentially connected bubblers cooled by a snow-water mixture. Gas temperature was measured by a thermocouple inserted 10–20 cm inside the vent.

Samples from the geothermal field are represented by steam vents (SV) and geothermal wells (W). Temperature of all steam vents was ~96°C corresponding to the boiling point at elevation of the geothermal field (800–900 m above sea level). Temperature of well fluids can be estimated between 250 and 300°C for the deep water-dominated wells and ~240°C for the shallow wells drilled into the steam cap. All wells are characterized by fluid enthalpies within a range of 900–2700 kJ/kg. Gas samples from geothermal wells were taken from the steam line of cyclone separators attached to each well. Separation pressure was near 8 atm everywhere.

In the laboratory a split of non-absorbed gases from the headspace was transferred into 10 ml ampoules for $\delta^{15}\text{N}$ and $^3\text{He}/^4\text{He}$ isotopic analyses. The headspace gases were analyzed by gas chromatography. Thermal conductivity detector, Ar as a carrier gas, and a packed column with Molecular Sieves 5A were used for determination of H_2 and He. Separation of Ar from O_2 was performed using a CT-III (Altech) packed column and He as a carrier gas. Same column was used for the analysis of N_2 , O_2 , CO, and CH_4 . Flame ionization detector and a packed column with Al_2O_3 were used for the analysis of light hydrocarbons C_2 – C_4 and CH_4 in low concentrations. Low concentrations of CO in low-temperature gases were determined using a packed MS 5A column coupled with a high-sensitivity Reduced Gas Detector (detection limit $\sim 10^{-4}\%$). Detection limits for N_2 , O_2 and Ar were 0.01, 0.01 and 0.005 vol%, respectively. For H_2 and He detection limit was estimated to be better than 0.001 vol%. Sensibility for methane was ~ 0.05 vol% with the thermal conductivity detector and better than 10^{-5}

vol% for the flame ionization detector. Sulfate (total sulfur after oxidation of the alkaline condensate by H_2O_2), chloride and fluoride were measured in the oxidized alkaline solution by ion chromatography. The CO_2 concentration was obtained by volumetric method after adding to an aliquot of the oxidized alkaline condensate an excess of concentrated H_2SO_4 and measuring the volume of the liberated CO_2 . Concentrations of SO_4^{2-} in selected samples were verified by the gravimetric determination of precipitated BaSO_4 . Uncertainties for the gas and chemical analyses were within 5%.

Condensates were analyzed for $\delta^{18}\text{O}$ and δD in the Institute of Ore Deposits (Moscow) and Instituto de Geofísica, UNAM, Mexico. The mass-spectrometric measurements in both laboratories were carried out by the DELTA Plus mass-spectrometer (Finnigan) coupled with the H/D device. Accuracy for $\delta^{18}\text{O}$ was about 0.2 ‰, for δD —about 0.5‰.

Helium and nitrogen isotopes were analysed in the Institute of Geophysics and Volcanology (INGV), Palermo. The $^3\text{He}/^4\text{He}$ ratios were measured by a static vacuum mass spectrometer (VG-5400TFT, VG Isotopes) modified to detect $^3\text{He}/^4\text{He}$ ion beams simultaneously, reducing the error of the $^3\text{He}/^4\text{He}$ measurements in ^3He -rich gases down to values lower than 0.1%. $^3\text{He}/^4\text{He}$ ratios were corrected for the atmospheric contamination on the basis of the difference between the $^4\text{He}/^{20}\text{Ne}$ of the sample and in the air (Sano and Wakita 1985). $\delta^{15}\text{N}$ measurements were carried out on a Finnigan Delta Plus XP continuous-flow IRMS coupled with a TRACE GC equipped with a MS 5A PLOT capillary column (60 m, ID 0.32 mm). The amount of ^{36}Ar , the isotopic composition and amount of nitrogen, have been analyzed directly and simultaneously on the same aliquot of a sample with a single injection (Inguaggiato et al. 2006). This procedure allows the correction of the air contamination and the normalization of the results. The analytical precision of $\delta^{15}\text{N}$ measurements is better than 0.15‰ (1σ) as determined on ten air samples.

Results and discussion

Gas output rates

On September the 1st 1999 Fischer (unpublished) has measured SO_2 flux of Mutnovsky with COSPEC. The flux was estimated to be 200 ± 50 tons/day. Assuming the weight-average SO_2 content in AF gases at 2.0% (Taran et al. 1992; this work) the total discharge will be of 10000 ± 2500 t/d or 115 ± 29 kg/s of volcanic vapor from the Active Funnel.

To estimate outputs from other fumarolic fields we used a comparative empirical method described by Hochstein

and Bromley (2001). The heat transferred to the atmosphere by fumarolic plumes is proportional to the visible plume area registered at the same time. Comparing areas of the reference combined plume of AF and fumarolic plumes of BF and UF fields, the next by magnitude output was accounted to the Upper Field at 20 ± 5 kg/s and only 6.6 ± 1.5 kg/s is emitted from the Bottom Field (Fig. 2b). The absolute values are 3–5 times lower than those from Vakin et al. (1976) and Polyak et al. (1985). However, we believe that the relative output from fumarolic fields ($\sim 0.8 : 0.15 : 0.05$) is estimated by this method with the acceptable accuracy.

Variations in chemical composition of volcanic and hydrothermal gases

Chemical and isotopic compositions of gases are listed in Tables 1, 2 and 3. Concentrations of N_2 and Ar are corrected on the direct air contamination using the measured O_2 content: $N_{2,c} = N_{2,m} - 3.72 \times O_2$ and $Ar_c = Ar_m - O_2/22.43$, where subscripts c and m stand for “corrected” and “measured”, respectively. Such a correction does not take into account O_2 consumed by the oxidation of sulfur species and hydrogen. AF fumaroles have the highest CO_2 -S- N_2 -He-Ar absolute concentrations and the lowest H_2O . Gases from the Bottom Field have a higher water content compared to the Active Funnel. The UF gases show an interesting combination of a relatively high gas temperature up to $280^\circ C$ together with a very high concentration of water vapor, low concentrations of other components (1–2%) and a high absolute concentration of methane. Relative concentrations of N_2 , Ar and CH_4 , are shown in Fig. 3a. Important sources for these components are thought to be air-saturated water (ASW) and hydrothermal water-rock interaction (Giggenbach 1992a; Taran and Giggenbach 2003). All gas compositions are plotted here including gases from geothermal wells and hydrothermal steam vents. Very low relative CH_4 content and N_2/Ar ratios much higher than air-ASW values are characteristic for all AF fumaroles, independently of the temperature. Hydrothermal gases show a mixing between CH_4 -poor and CH_4 -enriched components both associated with a meteoric-hydrothermal source. The highest relative (and absolute) concentration of methane is observed in the Upper Field gases. These gases are also characterized by elevated, non-atmospheric, N_2/Ar ratios and their compositions apparently lie on a trend between a high N_2/Ar , low CH_4 magmatic component and defined above a CH_4 -enriched meteoric-hydrothermal component. Another ternary plot for the “acidic” components CO_2 -S-Cl derived mostly from the degassing magma is shown in Fig. 3b for gases from the crater. It follows from this diagram that the AF gases have relatively constant HCl concentrations and variable CO_2 and S with a negative

correlation between them. The BF gases have the highest relative HCl concentration, and the UF gases are characterized by the lowest relative S and HCl.

The Giggenbach N_2 -Ar-He diagram (Giggenbach 1992a) shown in Fig. 3c for all gases, hydrothermal and volcanic, discriminates hydrothermal gases with a low He relative content and atmospheric N_2/Ar ratios, and high N_2/Ar volcanic gases with non-atmospheric (magmatic?) nitrogen. Gases from the Active Funnel with the highest N_2/Ar ratio (and hence, the highest fraction of the non-atmospheric nitrogen) are characterized by a high N_2/He ratio of 2000–3000.

Isotopic composition and origin of water

The isotopic data for rain and snowmelt water at the Mutnovsky area range in $\delta^{18}O$ from -19.1‰ to -10.7‰ and in δD from -124‰ to -72.6‰ (Taran et al. 1987; Cheshko and Esikov 1990). On the Craig plot, δD vs $\delta^{18}O$, (Fig. 4a) the meteoric water compositions are shown as a linear trend that represents the local meteoric water line (LMWL) for Kamchatka (Cheshko and Esikov 1990). This trend can be fitted with an expression $\delta D = (8 \pm 0.2) \cdot \delta^{18}O + 16.5 \pm 1.5$. Thus, the local meteoric water line is shifted approximately 6‰ above the global meteoric water line ($\delta D = 8\delta^{18}O + 10$, Craig 1961).

Only one point of the meteoric water composition is shown in Fig. 6 with $\delta D = -112\text{‰}$ and $\delta^{18}O = -15.7\text{‰}$ that corresponds to water from the crater lake formed after the 2000 phreatic eruption (Fig. 2) in the Southwest crater of Mutnovsky. The lake was filled by water from melting of perennial snow and ice accumulated inside the crater for many years (Zelenski et al. 2002). As the sample was taken soon after the lake had appeared, evaporation could not affect significantly its isotopic composition. Therefore this sample can be referred as the “average meteoric water” for the crater of the Mutnovsky volcano.

Hydrothermal vapors (geothermal wells and steam vents) have water isotopic composition close to meteoric water values. Water vapor from steam vents and geothermal wells (separators) both have a negative $\delta^{18}O$ shift of $\sim 1\text{‰}$ from the LMWL. As it has been shown by Taran et al. (1987), this negative O-shift associates with the steam separation at temperatures close to $200^\circ C$. At these temperatures the fractionation of hydrogen is close to zero but vapor is depleted in ^{18}O (Horita and Wesolowski 1994). In contrast, water from geothermal wells is ~ 1 – 1.5‰ enriched in ^{18}O . This O-shift due to water-rock interaction is smaller than for many other high-temperature hydrothermal systems indicating a high water-rock ratio in the Mutnovsky aquifer which is in turn in agreement with a very diluted character of the deep hydrothermal water (~ 200 ppm of Cl, Taran et al. 1986, 1987).

Table 1 Chemical composition (mmol/mol in the total discharge) and water isotopes (per mil vs. V-SMOW) of volcanic and hydrothermal gases from fumaroles in the crater of Mutnovsky volcano and steam vents and geothermal wells of Mutnovsky geothermal field. bd – below detection limit; empty cell – not determined. N₂ and Ar are corrected on the air contamination (see text). All geothermal wells were sampled several times during 2006. The data for wells are mean values over 4–5 analyses

NN°	Date	t°C	δ ¹⁸ O	δD	S	HF	HCl	CO ₂	H ₂ O	NH ₃	He	H ₂	CO	Ar _c	N _{2c}	CH ₄	N ₂ /He	N ₂ /Ar	
Active funnel																			
1	09.2001	507	-3.5		11.49		2.03	14.08	973.0		7.98E-05	0.309	0.002431	0.00022	0.199	0.000108	2494	905	
2	09.2001	196	-3.2		12.00		0.92	9.23	977.9		5.74E-05	0.002	0.000005		0.098	0.000119	1708		
3	09.2001	309	-4.6		7.01		1.80	14.21	977.9		9.26E-05	0.008	0.000007	0.00103	0.313	0.000107	3382	304	
4	09.2001	240	-0.9		17.62		2.38	11.22	969.9		6.76E-05	0.008	0.000005	0.000126	0.139	5.36E-05	2056	1103	
5	09.2001	410	-2.3		12.35		2.85	14.86	971.6		8.58E-05	0.050	0.000127	0.00016	0.174	7.6E-05	2028	1088	
6	09.2001	450	1.0		18.39		2.54	10.76	969.4		6.66E-05	0.131	0.000229	0.00089	0.157	4.8E-05	2357	176	
7	09.2001	383	-4.1		10.76		1.70	16.69	971.6		0.000107	0.034	5.97E-05	0.00109	0.379	0.000152	3533	348	
8	09.2007	425	-2.1	-49.5	1.03	0.87	0.07	5.68	993.1		0.000139	0.125	0.00060	0.00030	0.349	0.00015	2518	1160	
9	09.2007	380	-0.7	-58.4	22.83	0.34	1.81	6.09	968.7		6.65E-05	0.036	0.00002	0.00025	0.141	0.00002	2122	566	
10	09.2007	543	-1.4	-53.5	12.07	0.31	2.35	23.72	960.3	0.033	0.000152	0.716	0.01228	0.00041	0.467	0.00048	3080	1138	
11	09.2007	594	-1.1	-54.5	13.16	0.36	2.38	26.04	956.8	0.010	0.000162	0.779	0.01042	0.00040	0.443	0.00034	2737	1116	
12	09.2007	590	0.5	-45.3	12.69	1.07	2.33	24.65	958.2		0.00016	0.664	0.00632	0.00019	0.369	0.00022	2304	1967	
13	09.2007	580	-1	-46.5	14.08	0.74	2.46	19.72	967.1	0.030		0.527	0.00378	0.00025	0.346	0.00007		1393	
14	09.2007	315	-0.65	-55.2	20.21	0.22	2.02	11.26	966.1		8.69E-05	0.011	0.00002	0.00018	0.175	0.00005	2017	952	
15	09.2007	543	1.3	-42.7	17.58	1.00	2.72	16.06	965.9		0.00012	0.315	0.00144	0.00032	0.224	0.00007	1877	701	
16	09.2007	357	-2.5	-58.9	14.07	0.31	2.10	12.64	970.6		8.17E-05	0.030	6.26E-05	0.00019	0.224	0.00015	2748	1167	
17	09.2007	400	-4.2	-72.6	8.12	0.24	1.32	23.61	966.3		0.000144	0.056	7.26E-05	0.00118	0.332	0.00068	2300	281	
18	09.2007	205	-4.3	-79	15.25	0.09	0.78	2.78	981.0		4.96E-05	0.006	bd	0.00013	0.070	0.00049	1415	537	
19	09.2007	543	1.5	-41.1	17.58	1.00	2.80	16.56	965.9	0.005	0.00012	0.315	0.001438	0.00032	0.224	0.00007	1877	701	
20	09.2007	280	-0.4	-57.4	24.47	0.26	1.68	1.83	971.7		2.82E-05	0.006	bd	0.00011	0.029	0.00021	1029	263	
21	09.2007	380	-5.9	-73.4	10.28	0.31	2.27	21.14	965.6	0.017	0.000133	0.039	3.75E-05	0.00014	0.341	0.00086	2554	2458	
Bottom Field																			
22	09.2001	153	-5.9	nd	3.62	nd	2.21	5.12	988.5		8.7E-05	0.0085	7.53E-05	0.0008	0.14	0.002885	1610	175	
23	09.2007	111	-8.84	-97.6	2.94	0.01	0.45	2.85	993.7		9.32E-06	0.003	4.66E-06	0.00013	0.044	0.00279	4773	347	
24	09.2007	137	-5.17	-82.2	3.79		1.22	3.05	991.9	0.023	2.27E-05	0.002	1.41E-06	0.00018	0.064	0.00106	2802	351	
25	09.2007	263	-4.68	-86.4	3.67	0.07	1.92	2.58	991.7	0.011	2.6E-05	0.009	6.84E-07	0.00004	0.045	0.00297	1742	1038	
26	10.2006	120	-4.94	-77.4	6.73	0.01	2.63	5.71	984.8		5.23E-05	0.005	bd	0.00011	0.078	0.00307	1492	693	
27	10.2006	98	-6.79	-86.7	2.12		0.03	5.25	992.5		4.89E-05	0.004	1.44E-06	0.00014	0.083	0.00317	1694	596	
28	10.2006	186	-5.03	-76.3	2.91	0.01	1.17	1.72	994.1	0.011	4.82E-05	0.006	1.28E-06	0.00014	0.070	0.00332	1452	498	
29	10.2006	98	-5.43	-82.1	1.61		0.03	3.11	995.2		4.58E-05	0.005	2.21E-06	0.00014	0.069	0.00261	1512	485	
30	10.2006	105	-5.16	-76.6	3.26		0.91	2.10	993.6		5.18E-05	0.002	1.32E-06	0.00012	0.080	0.00187	1544	643	
31	09.2007	125					5.33	994.6			7.56E-06	0.003	bd	0.00013	0.036	0.00127	4806	272	
32	09.2007	218					3.38	996.6			1.38E-05	0.002	1.15E-06	0.00007	0.029	0.00120	2086	411	
33	10.2007	190					4.53	995.4			3.1E-05	0.005	2.32E-06	0.00024	0.073	0.00280	2369	303	
34	10.2007	150					4.75	994.9			0.000122	0.002	bd	0.00033	0.325	0.00205	2653	98	
35	10.2007	115					3.73	996.2			3.87E-05	0.003	1.07E-06	0.00030	0.074	0.00184	1901	246	
36	10.2007	135					0.10	999.9			4.37E-05	0.001	1.21E-06	0.00029	0.079	0.00272	1796	274	

37	10.2007	150		3.95	995.8	4.49E-05	0.002	5.93E-05	0.00201	0.220	0.00231	4909	110
Upper field													
38	2001	281	-11.3	0.07	993.1	1.12E-05	0.004996	6.34E-06	0.01442	0.091	0.03225	8159	188
39	09.2007	96	-13.7	0.02	995.1	6.39E-06	0.002	bd	0.00018	1.175	0.02407	5346	276
40	09.2007	138	-9.5	0.13	993.2	7.04E-06	0.001	bd	0.00018	0.034	0.01474	7057	188
41	09.2007	98	-11.2	0.22	997.6	8.28E-06	0.001	bd	0.00024	0.050	0.01978	5520	248
42	09.2007	186	-11.4	0.01	996.7	1.42E-05	0.004	8.4E-07	0.00013	0.046	0.02205	2295	274
43	10.2006	149	-11.0	0.06	997.4	2.03E-05	0.003	8.88E-07	0.00014	0.033	0.02395	1915	199
44	10.2006	212	-11.6	0.07	994.0	0.032	0.003	9.49E-07	0.00019	0.039	0.02281	3901	270
45	10.2006	285	-11.6	0.09	997.4	0.063	0.007	bd	0.00014	0.037	0.02503	3256	244
46	10.2006	148	-11.7	0.12	996.6	1.12E-05	0.001	9.75E-07	0.00015	0.036	0.02739	2697	203
47	10.2006	231	-11.7	0.08	996.9	1.35E-05	0.005	9.31E-07	0.00019	0.039	0.02028	3904	258
48	10.2006	248	-11.2	0.08	995.2	0.059	0.007	8.72E-07	0.00013	0.035	0.02239	2885	
49	10.2006	172	-11.9	0.07	996.7	1.21E-05	0.005	bd	0.00035	0.024	0.01392		49
Steam vents													
50	05.2007	96			999.6		0.179	1.02E-05	0.00119	0.058	0.01730		60
51	05.2007	96	-16.6	0.01	999.1	7.15E-06	0.444	bd	0.00163	0.098	0.01562	13719	56
52	05.2007	96	-15.5	0.00	998.0	9.82E-06	0.140	3.5E-05	0.00120	0.067	0.00346	6871	52
53	05.2007	96		0.00	999.6		0.258	7.51E-06	0.00182	0.095	0.01094		69
54	05.2007	96	-14.5	0.02	999.0	1.93E-06	0.043	bd	0.00021	0.015	0.00387	7515	68
55	05.2007	96		0.00	998.0		0.164	bd	0.00035	0.024	0.01392		
56	05.2007	96		0.01	998.4								
57	05.2007	96		0.00	996.8								
58	05.2007	96		0.05	999.9								
Geothermal wells													
59	24	910	-14.0	0.22	999.70	6.50E-07	0.001	2.50E-07	0.00006	0.030	0.00096	10769	49
60	048	1105	-13.5	0.32	999.60	8.79E-07	0.001	5.17E-07	0.00012	0.106	0.00026	12059	86
61	049	1018	-13.2	0.88	998.98	1.79E-06	0.005	3.77E-06	0.00056	0.0394	0.00145	21970	70
62	013	1425	-13.4	0.57	999.24	1.15E-06	0.023	6.01E-06	0.00015	0.150	0.00168	13034	99
63	26	2650	-14.2	1.00	998.80	4.15E-06	0.027	bd	0.00090	0.403	0.00182	9699	45
64	A-2	1293		0.38	999.48	2.85E-07	0.006	7.68E-07	0.00004	0.024	0.00037	8361	54
65	029W	1297	-13.2	0.45	999.41	2.85E-07	0.015	bd	0.00005	0.023	0.00044	7924	42
66	GK-1	1166	-14.1	0.43	999.52	3.85E-07	0.027	1.93E-06	0.00009	0.045	0.00126	11675	50
67	016	2640		0.97	998.76	1.18E-06	0.045	1.69E-06	0.00021	0.109	0.00188	9277	52
68	037	1519	-13.2	0.65	999.23	1.18E-06	0.006	2.36E-06	0.00013	0.055	0.00327	4691	43
69	042	1113		0.41	999.46	1.12E-06	0.001	1.66E-06	0.00042	0.0210	0.00100	18729	50
70	4E	1153		0.36	999.54	8.79E-07	0.001	1.4E-06	0.00027	0.163	0.00072	18534	60
71	053	1108		0.25	999.62	3.81E-06	0.002	bd	0.00048	0.0327	0.00070	8584	68

Table 2 Isotopic compositions of He (R/R_a, where R is ³He/⁴He in a sample and R_a=1.4×10⁻⁶—isotopic composition of He in air) and N₂ (δ¹⁵N relative air) in volcanic and hydrothermal gases of Mutnovsky. Helium ratios are corrected on air contamination using ⁴He/²⁰Ne. Excess N₂ is calculated from N₂/³⁶Ar as (1-r_a/r)×100%, where r and r_a are N₂/³⁶Ar in a sample and in air (asw), respectively. NH₃/N₂ ratios are done as mean values for each group of samples

Date	R/R _a	He/Ne	δ ¹⁵ N	N ₂ / ³⁶ Ar	N ₂ ,ex % (asw/air)	T°C	NH ₃ /N ₂
Active funnel							
09/2007	5.29	38	3.3	186420	94/87	543	
09/2007	5.52	67	1.6	58081	81/58	594	
09/2007	6.58	28	1.9	63250	82/61	315	
09/2007	5.57	35	2.5	123077	91/80	543	<0.05>
09/2007	5.99	79	2.8	174652	94/86	400	
10/2007	6.58	13	1.4	39941	72/38	315	
09/2007	5.83	4.3	0.24	29241	62/16	590	
Bottom field							
10/2007	7.59	71				137	
10/2007	7.67	126	0.16	77710	86/68	263	<0.3>
10/2003	7.36	6.1				96	
Upper field							
05/2006	6.98	31	0.57	52580	77/53	285	
05/2006	6.59	36	-1.0	26578	59/7.2	248	<1.0>
Steam vents							
23/10/07	6.06	4.9	-0.08	14710	24/-68	96	
23/10/07	6.66	5.1	-0.15	18592	40/-33	96	<1.0>
23/10/07	5.47	4.3	0.23	16728	33/-47	96	
Geothermal wells							
23/10/07	3.51	25	-1.80	16674	33/-47	024	
23/10/07	3.97	14	0.30	17819	37/-38	049	<0.8>
23/10/07	5.89	15	-0.33	19517	43/-26	A2	
23/10/07	3.99	3.0	0.13	20651	46/-19	013	
23/11/07	5.77	3.0	-3.4	15452	27/-60	045	
Bubbling gas							
23/11/07	5.97	2.1				45	
23/11/07	6.53	5.0				60	
23/11/07	5.13	1.2				47	
air	1.0	0.29	0.0	24660			
asw	1.0	0.25	0.0	11210			

Isotopic composition of volcanic vapors from crater fumaroles (Fig. 4a) is typical for the subduction type volcanoes (Sakai and Matsubaya 1977; Taran et al. 1989; Giggenbach 1992b). Most of the δD-δ¹⁸O data points lay close to the mixing trend between slightly positively O-shifted meteoric waters and “andesitic” waters (Taran et al. 1989). “Andesitic” (Taran et al. 1989) or “arc-magmatic” (Giggenbach 1992b) water is thought to be the subducted and partially altered seawater releasing through arc volcanoes. The δD-HCl relationship shown in Fig. 4b is also similar to that observed for other arc volcanoes with high-temperature, “magmatic”, fumaroles (Taran et al. 1995) indicating a common source of Cl and water enriched in D.

The Upper Field fumaroles heated up to 280°C discharge almost pure meteoric steam with a very low chloride and a narrow range of isotopic composition: δD=-106±2‰, δ¹⁸O=-11.5±0.5‰. The steam is slightly ¹⁸O-shifted (Fig. 4a). A shallow mixing with the ice melt water from

glacier for this, closest to the glacier, fumarolic field, seems to be negligible. A significant scattering of the Bottom Field points may indicate some more complicated processes of steam-brine separation beneath the crater floor as mentioned by Taran et al. (1992).

According to Bindeman et al. (2004), a value of +5.9±0.5‰ can be accepted for δ¹⁸O of lavas of Mutnovsky. Extrapolation of the mixing line to this value gives for δD of the parent magmatic fluid of Mutnovsky a value of -19‰.

Fractions of magmatic fluid for most important fumarole groups can be calculated according to their relative positions on the mixing trend (Figs. 4a and 12b). Gas from the most powerful group of fumaroles in the Active Funnel, which is responsible for at least 80% of the total emission of the volcano, contains 61±3% of magmatic fraction. At the same time, the second by magnitude Upper Field (~15% of the total emission) discharge gas with not more than 2% of magmatic fluid.

Table 3 Light hydrocarbons in volcanic and hydrothermal fluids of Mutnovsky. Concentrations are normalized to methane ($c_i = C_i/C_{CH_4}$). For absolute concentrations of methane see Table 1

NN°	CH ₄	C ₂ H ₆	C ₂ H ₄	C ₃ H ₈	C ₃ H ₆	C ₄ H ₁₀	T°C
				Active	funnel		
8	1	0.0024	0.0054	2.8E-04	0.0065	1.9E-04	425
9	1	0.0022	6.1E-04	5.1E-04	2.4E-04	2.3E-04	543
10	1	0.0024	0.0044	0.0022	0.0016	7.0E-04	543
11	1	0.0026	7.1E-04	1.3E-04	0.0019	bd	594
12	1	0.0023	0.0032	7.9E-04	0.0016	5.0E-04	590
13	1	0.0020	0.0036	4.0E-04	0.0022	2.9E-04	580
14	1	0.0025	8.9E-04	3.8E-04	9.0E-04	2.9E-04	315
15	1	0.0027	0.0062	6.9E-04	0.0038	4.3E-04	543
16	1	0.0022	0.0011	3.0E-04	8.7E-04	1.2E-04	357
17	1	bd	1.1E-04	2.3E-05	1.4E-04	2.5E-05	400
18	1	0.0025	2.8E-04	1.4E-04	7.1E-05	1.0E-04	205
19	1	0.0027	0.0062	6.9E-04	0.0038	4.3E-04	543
20	1	0.0044	0.0012	5.0E-04	0.0011	1.4E-04	280
21	1	0.0020	3.5E-04	1.7E-04	1.9E-04	4.0E-05	380
				Bottom	field		
22	1	0.0013	4.2E-05	3.0E-05	1.4E-05	2.3E-05	111
23	1	0.011	7.8E-05	1.3E-05	9.8E-05	1.1E-05	137
24	1	0.017	0.012	1.0E-05	7.6E-06	7.3E-06	263
25	1	0.015	bd	3.3E-06	6.7E-06	6.2E-06	98
26	1	0.026	bd	1.1E-04	2.4E-05	2.6E-05	186
27	1	0.016	bd	6.2E-06	2.2E-05	6.7E-06	98
28	1	0.015	bd	1.6E-05	2.9E-05	4.4E-04	105
29	1	0.026	4.5E-05	4.3E-05	1.2E-05	1.4E-04	125
30	1	0.017	7.2E-04	8.5E-06	1.1E-06	6.5E-04	218
31	1	0.025	8.2E-05	2.0E-05	9.1E-06	2.2E-04	190
32	1	0.011	0.0038	6.8E-05	6.7E-05	2.4E-04	150
33	1	0.019	2.0E-04	4.9E-05	9.0E-06	2.5E-04	115
34	1	0.016	1.3E-04	1.5E-05	2.9E-06	3.5E-06	135
35	1	0.0092	9.0E-05	1.5E-05	1.3E-05	1.1E-05	150
				Upper	field		
38	1	0.0063	bd	2.3E-04	3.9E-06	5.6E-06	96
39	1	0.0087	bd	2.5E-04	4.2E-06	3.3E-06	138
40	1	0.0085	bd	3.7E-04	3.5E-06	2.8E-06	98
41	1	0.0078	bd	3.7E-04	4.9E-06	1.6E-06	186
42	1	0.011	bd	2.9E-04	6.9E-07	1.6E-06	149
43	1	0.011	bd	3.4E-04	5.7E-06	3.4E-06	212
44	1	0.014	0.0093	3.3E-04	6.0E-06	3.2E-06	285
45	1	0.011	bd	3.0E-04	7.6E-07	1.5E-06	148
46	1	0.011	0.0096	3.3E-04	6.5E-06	1.9E-06	231
47	1	0.013	0.0053	3.6E-04	5.8E-06	3.0E-06	248
48	1	0.015	bd	3.2E-04	6.7E-06	4.7E-06	172
				Steam	vents		
50	1	0.013	4.0E-04	7.1E-04	1.8E-05	6.9E-05	96
51	1	0.020	1.6E-04	4.0E-04	4.4E-05	8.6E-05	96
52	1	0.030	5.1E-05	1.9E-03	7.8E-05	2.1E-04	96
53	1	0.028	5.5E-05	1.7E-03	5.2E-05	1.9E-04	96
54	1	0.0046	9.3E-05	5.7E-04	3.9E-05	9.6E-05	96

Table 3 (continued)

NN°	CH ₄	C ₂ H ₆	C ₂ H ₄	C ₃ H ₈	C ₃ H ₆	C ₄ H ₁₀	T°C
55	1	0.035	7.1E-05	4.0E-03	3.0E-05	2.2E-04	96
				Geothermal	wells		
59	1	0.0021	1.5E-04	6.2E-05	1.3E-05	3.4E-05	
60	1	0.024	4.6E-04	0.016	5.3E-04	6.6E-04	
61	1	0.026	3.2E-04	0.0018	6.8E-04	2.6E-04	
62	1	0.0075	0.0014	8.8E-04	1.5E-05	2.6E-05	
63	1	0.024	1.3E-04	0.0019	1.9E-04	4.8E-04	
64	1	0.014	3.3E-04	0.0012	8.9E-05	2.7E-04	
65	1	0.021	1.4E-04	0.0019	2.9E-04	6.3E-04	
66	1	0.019	2.2E-04	0.0011	6.9E-05	2.2E-04	
67	1	0.014	5.9E-05	6.4E-04	2.1E-06	2.2E-05	
68	1	0.021	5.1E-04	0.0019	2.6E-04	8.8E-04	

nd not determined; bd below detection limit. Numbers of samples according to Table 1

Major and minor reactive species and chemical equilibria

Hydrogen and carbon monoxide

Many of the chemical and isotopic features observed in volcanic gases can be explained in terms of the attainment of chemical equilibrium among the components of the gas phase itself or within the heterogeneous system involving the gas-rock interaction. Giggenbach (1987) developed a method of geothermometers and geobarometers for volcanic gases using a set of redox-equilibria with $R_H = \log(x_{H_2}/x_{H_2O})$ as the main variable. According to Giggenbach (1987), the redox-state of high-temperature volcanic gases is controlled by fast reactions between oxidized and reduced sulfur species in the gas phase itself. With decreasing temperature the analytical R_H preserve high values due to quenching of high-temperature values or attaining equilibrium with the rock matrix. The R_H vs temperature diagram for Mutnovsky gases (Fig. 5a) demonstrates a good correspondence of the measured gas compositions in the Active Funnel to the “gas-buffer” control for gases with temperature $>300^\circ\text{C}$ and a systematic deviation to more “hydrothermal” values as temperature decreases. Another system which was found to adjust rapidly to conditions into the gas phase itself is CO-CO₂ (Giggenbach 1987; Taran et al. 1995; Chiodini and Marini 1998). Comparison of analytical and theoretical CO/CO₂ ratios, as shown in Fig. 5b, reveals the same behavior as for R_H : at temperatures $>300^\circ\text{C}$ the relative concentration of CO is governed by the gas (H₂S-SO₂) buffer and at lower temperature the deviation of data points from the gas buffer line can be related either to quenching of the high-temperature values or to re-equilibration with rock (FeO_{1.5}-FeO, or GT-buffer—from Geothermal Buffer) as defined by Giggenbach (1987).

Points for geothermal steam vents lay close to the GT-vapor buffer line, whereas all points for geothermal wells fall into the two-phase region, between GT-liquid and GT-vapor lines.

Methane

Methane in volcanic and hydrothermal gases of Mutnovsky is not in equilibrium within the C-H-O system, similar to observed for many volcano-hydrothermal systems (Taran 1986; Taran and Giggenbach 2003). Three plots “log(CH₄/CO₂) vs R_H ” with the same set of data points and different theoretical equilibrium lines are shown in Fig. 6. Line GT corresponds to the FeO-FeO_{1.5} rock buffer. The closest to an equilibrium line data points correspond to samples from the highest-temperature ($\sim 600^\circ\text{C}$) AF fumaroles. It could be suggested that CH₄ was equilibrated in the fluid under ~ 1 kbar pressure. Taran and Giggenbach (2003) have discussed in details processes potentially responsible for the equilibration of hydrocarbons in volcanic and hydrothermal fluids. Since methane and other hydrocarbons in volcanic and hydrothermal systems have predominantly sedimentary origin, the main process leading to the equilibration should be oxidation of methane (decreasing concentrations with increasing temperature) but not reduction of CO₂ because this process even under appropriate reduced conditions is extremely slow kinetically (Giggenbach 1997a; Taran and Giggenbach 2003). Therefore, the oxidation (oxidative hydrolysis or cracking) of hydrothermal methane admixed to the high-temperature AF gases may be responsible for much lower CH₄ concentrations in AF gases.

Hydrocarbons

The histogram in Fig. 7 characterizes the hydrocarbon fraction of volcanic and hydrothermal gases in terms of

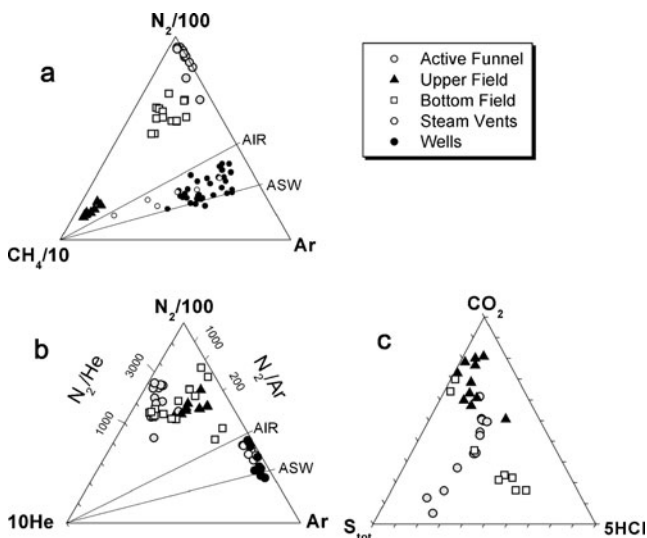


Fig. 3 Ternary diagrams for Mutnovsky volcanic and hydrothermal gases: **a:** N₂-CH₄-Ar; **b:** N₂-Ar-He, and **c:** S_{tot}-CO₂-HCl (mmol/mol basis). See text for discussion

their CH₄/C₂₊ ratios where C₂₊ is the sum of concentrations of all C₂ to C₄ hydrocarbons. The mean value of the CH₄/C₂₊ ratio of ~60 for all but the Active Funnel hydrocarbons indicates essentially thermogenic origin of hydrocarbons, typical for high-temperature hydrothermal systems and volcanic gases (Welhan 1988; Taran and Giggenbach 2003). Taran et al. (1987, 1992) reported isotopic composition of CH₄ in volcanic and hydrothermal gases of Mutnovsky in the range of -25‰ to -27‰, typical for thermogenic methane.

Much higher CH₄/C₂₊ ratios in the AF hot gases most probably is the result of high-temperature hydrolytic oxidation of C₂₊ hydrocarbons, which is more effective for long hydrocarbon chains than for methane (Arutyunov

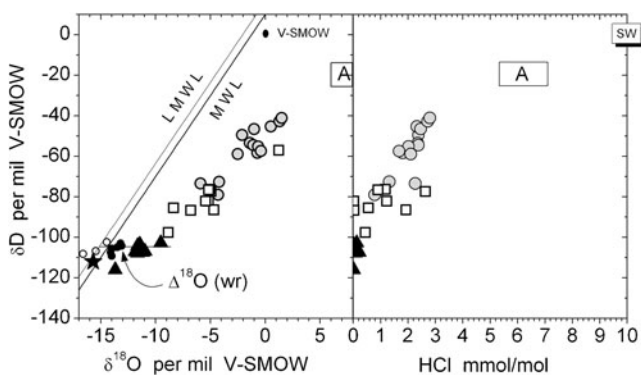


Fig. 4 δD vs δ¹⁸O plot for volcanic and hydrothermal vapors of Mutnovsky. For symbols see Fig. 3. LMWL—local meteoric water line—is plotted using data by Cheshko and Esikov (1990) MWL—world meteoric water line, 8δ¹⁸O + 10 (Craig, 1961). Areas A corresponds to “Arc” or “Andesitic” waters as defined by Taran et al. (1989) and Giggenbach (1992b). SW—seawater. Δ¹⁸O (wr) denotes the oxygen isotopic shift due to water-rock interaction (see text)

and Krylov 1998). The only chemical processes that can lead to equilibrium relationships between C₂₊ hydrocarbons are those without breaking or forming the C-C bonds. The most probable are hydrogenation-dehydrogenation of an alkane-alkene pair like C₂H₄ + H₂ = C₂H₆. Taran and Giggenbach (2004) showed that in volcanic gases of White Island the metastable equilibrium is attained between C₂H₆ and C₂H₄ as well as between C₃H₈ and C₃H₆. Gases of Mutnovsky volcano and hydrothermal system show that metastable equilibrium between alkanes and alkenes is attained at temperatures close to the discharge temperatures. Data points in Fig. 8 fall between the equilibrium line at 1 bar (surface) pressure and the line corresponding to equilibrium under pressure of saturated vapor from a hypothetical “brine” as it has been suggested by Giggenbach (1987) for quantifying the temperature-pressure relationship beneath the White Island crater.

He as an indicator of the parent magmatic gas composition

Balance calculations concerning the subducted and the releasing through volcanoes volatiles show that more than 80% of He is derived from the mantle wedge and the contribution of volatiles from the subducting slab to He budget is low (Fischer et al. 1998; Hilton et al. 2002; Taran 2009). On the other hand, CO₂ and N₂, S, Cl and H₂O in arc magmas originate predominantly from the slab, i.e. from sediments and altered oceanic crust (AOC). All volatiles become coupled in the magma-generation zone and rise to the surface altogether in most cases as dissolved in magmas. Magma degassing in magma chambers or magmatic conduits may disturb the initial volatile composition due to different solubilities in magmas (Carroll and Webster 1994; Nuccio and Paonita 2001). N₂, Ar and He have low solubilities in silicate melts, comparable within one order of magnitude, (Nuccio and Paonita 2001; Miyazaki et al. 2004) and have low (N₂) or no (Ar, He) reactivity. Therefore they remain almost coupled until the surface. CO₂ has also low solubility in magmas, lower than that of HCl, H₂O and sulfur (Carroll and Webster 1994). Moreover, these components have also low solubility in water and therefore will not be affected by the condensation of volcanic vapor at shallow levels. If the predominant source of He, CO₂, Ar and N₂ is magma, the corresponding volcanic gases must demonstrate a simple mixing trend between almost zero-He, Ar, N₂ and CO₂ air-saturated water vapor and a parent magmatic gas composition. A set of diagrams with different species plotted vs He is shown in Fig. 9. Very good positive correlations with He can be seen for CO₂ and N₂ and not so good but still doubtless for other components except the total sulfur. CO₂-He and N₂-He data points show a mixing between almost zero-N₂ and CO₂ meteoric endmember and an endmember with N₂/He close to 2400 and CO₂/He ~1.6×

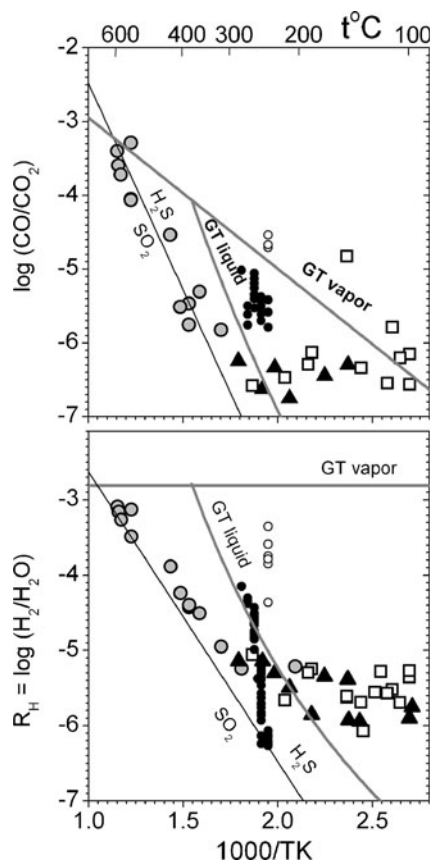


Fig. 5 Redox-diagrams for gases from the Mutnovsky crater in terms of $\log(\text{CO}/\text{CO}_2)$ vs temperature and $R_H = \log(X_{\text{H}_2}/X_{\text{H}_2\text{O}})$ vs temperature. Lines for the “rock” buffer (GT vapor and GT liquid) and “gas” buffer (H_2S - SO_2) are plotted according to Giggenbach (1987, 1992). For symbols see Fig. 3

10^5 . Despite Cl, δD and $\delta^{18}\text{O}$ can be affected by deep and shallow processes of magma degassing, dissolution in groundwater and elemental and isotopic fractionation in processes of phase separation (crystallization, boiling), their correlation with He also indicates that a high He concentration in the volcanic gas is associated with a high Cl content and the enrichment of water vapor with heavy isotopes. There is no correlation of He with sulfur, as well as sulfur with other species (except CO_2 in AF gases, Fig. 4), indicating a high mobility of sulfur in shallow processes like precipitation of native S and remobilization from previous precipitates and incrustations in fumarolic conduits (Giggenbach 1987; Taran et al. 1992; Chiodini et al. 1993).

The main feature of volcanic gases of Mutnovsky is that three separated in space fumarolic fields located within the same volcano edifice discharge gases with different and apparently “incompatible” compositions—they do not lie on single mixing trends. The Upper Field fumaroles, in spite of their relatively high temperature (up to 280°C), discharge almost pure hydrothermal fluid with a very low fraction of magmatic components. The Bottom Field fumaroles probably

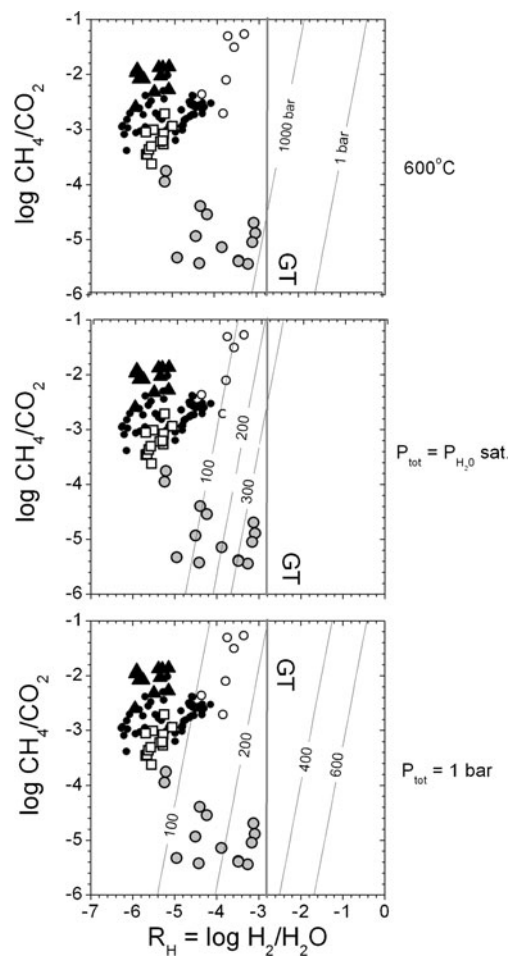


Fig. 6 Variations in $X_{\text{CH}_4}/X_{\text{CO}_2}$ in volcanic and hydrothermal gases of Mutnovsky as a function of discharge temperature. Lines of full equilibrium for different pressures at 600°C (up); Equilibrium lines for different temperatures at saturation pressure (middle); Equilibrium lines for different temperatures at 1 atm pressure. GT line ($R_H = -2.81$) corresponds to the “rock” buffer (Giggenbach, 1987; Taran and Giggenbach, 2003). For symbols see Fig. 3

affected by shallow boiling and condensation processes providing high variations in the chemical compositions of fluids. Though they apparently are a link between “hydrothermal” UF vents and “magmatic” AF fumaroles, their data points may have distinct slopes on N_2 -He and CO_2 -He plots (Fig. 9).

Mixing relationships for CO_2 , N_2 and He for AF fumaroles that discharge more than 80% of the total gas output from Mutnovsky may be interpreted as the two-endmember mixing of meteoric-hydrothermal vapor and a magmatic component from a single source with CO_2/He and N_2/He ratios estimated from the slopes of the corresponding mixing trends. The Bottom Field fumaroles may be a mixture of the same meteoric hydrothermal component but with a fluid from another magmatic source. Active Funnel and Bottom Field may release gases from different, separated in space, magma chambers with different magma composition or different degree of degassing. It is supported also by

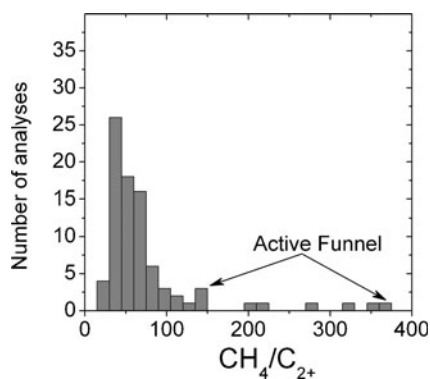


Fig. 7 Histogram showing distribution of $\text{CH}_4/\text{C}_{2+}$ ratios in volcanic and hydrothermal gases of Mutnovsky. Mean value of 61 correspond to “thermogenic” origin of hydrocarbons. See text for discussion

the distinct N_2 -Ar and He-Ar correlations shown in Fig. 10. The magmatic fluid source for the Active Funnel has $\text{N}_2/\text{He} \sim 2400$; $\text{N}_2/\text{Ar} \sim 1000$ and $\text{He}/\text{Ar} \sim 0.4$. The Bottom Field fumaroles are fed from a source with $\text{N}_2/\text{He} \sim 1760$; $\text{N}_2/\text{Ar} \sim 370$ and $\text{He}/\text{Ar} \sim 0.2$. The Upper Field gases are probably a mixture between BF mixed fumarolic gas and the same meteoric-hydrothermal fluid. CO_2 in BF fumaroles (Fig. 9) varies independently of He with large variations of CO_2/He . This can be caused by mixing of BF gases with a random amount of meteoric-hydrothermal fluid which has relatively high concentrations of CO_2 , variable concentrations of N_2 and very low concentrations of He.

The parent magmatic gas

Good correlations between He and major species in AF hot gases may be used for the estimation of the composition of the parent magmatic gas feeding volcanic gases from AF. As it was outlined above, the isotopic composition of the parent gas may be approximated as $\delta\text{D} = -19\text{‰}$ and $\delta^{18}\text{O} = +5.9\text{‰}$ (see Fig. 11 for details). The latter value is the main reference point based on the rock isotopic composition (Bindeman et al. 2004) and an assumption that magmatic water has the same oxygen isotopic composition as the parent magmatic melt. The procedure of estimation of the composition of parent gas is shown in Fig. 12. He concentration in the magmatic component can be determined as 0.00022 mmol/mol from linear regression of the δD -He and $\delta^{18}\text{O}$ -He correlations (Fig. 11b). Then, concentrations of H_2O , CO_2 , N_2 and HCl can be determined by linear regression of the corresponding mixing relationships (Fig. 12). The composition of the parent gas is presented in Table 4 together with average compositions of fluids from BF and UF fields, steam vents and geothermal wells. Each component is done as a mean value with a corresponding standard deviation (1σ). Sulfur in the parent gas was calculated from the mass balance by the difference $1000 - \sum C_i$ (mmol), where C_i are the estimated concentrations of other major components.

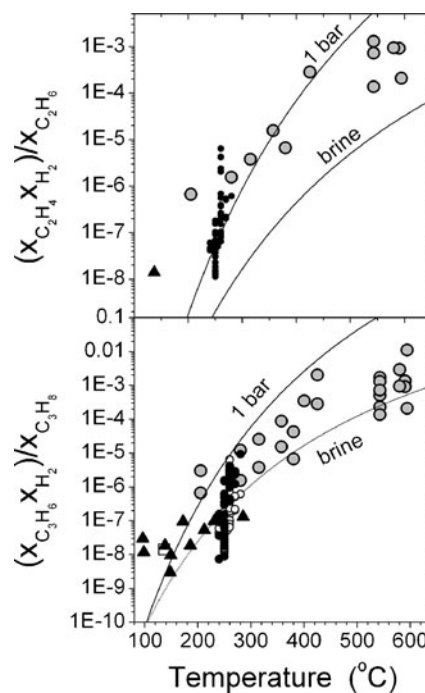


Fig. 8 Variations in the quotient ratios for reactions $\text{C}_2\text{H}_4 + \text{H}_2 = \text{C}_2\text{H}_6$ (upper panel) and $\text{C}_3\text{H}_6 + \text{H}_2 = \text{C}_3\text{H}_8$ (lower panel) as a function of discharge temperature for Mutnovsky volcanic and hydrothermal gases. The 1 bar equilibrium line corresponds to metastable equilibration under surface conditions. The “brine” line corresponds to vapors derived from a hypothetical saline brine with pressure controlled by the boiling of such a brine (Giggenbach 1987). For symbols see Fig. 3

The Mutnovsky parent gas composition is different from the Kudryavy parent gas estimated by Taran et al. (1995) and Fischer et al. (1998) in H_2O , He, CO_2 and N_2 concentrations and almost identical to Kudryavy gas in S, Cl and Ar.

He isotopes in volcanic and hydrothermal gases of Mutnovsky

Three distinct fluid sources for fumaroles of Mutnovsky volcano (meteoric-hydrothermal, AF-source and BF-source) are in agreement with the presented and previous data on He-isotopes (Rozhkov and Verkhovsky 1990; Taran et al. 1992). Points with the highest absolute He concentration on the plot ${}^3\text{He}/{}^4\text{He}$ vs He concentration (Fig. 13) belong to the hottest AF gases and have ${}^3\text{He}/{}^4\text{He}$ values $< 6.6\text{Ra}$, where $\text{Ra} = 1.39 \times 10^{-6}$, the air ratio. However, the highest ${}^3\text{He}/{}^4\text{He}$ of $\sim 7.67\text{Ra}$ was observed in the Bottom Field fumaroles and the lowest values were measured in the Upper Field gases and fluids from hydrothermal vents and wells. Such a difference in He-isotope ratios for volcanic gases discharging from the same crater is difficult to explain taking into account that the Bottom Field gases, according to their chemical and water isotope compositions,

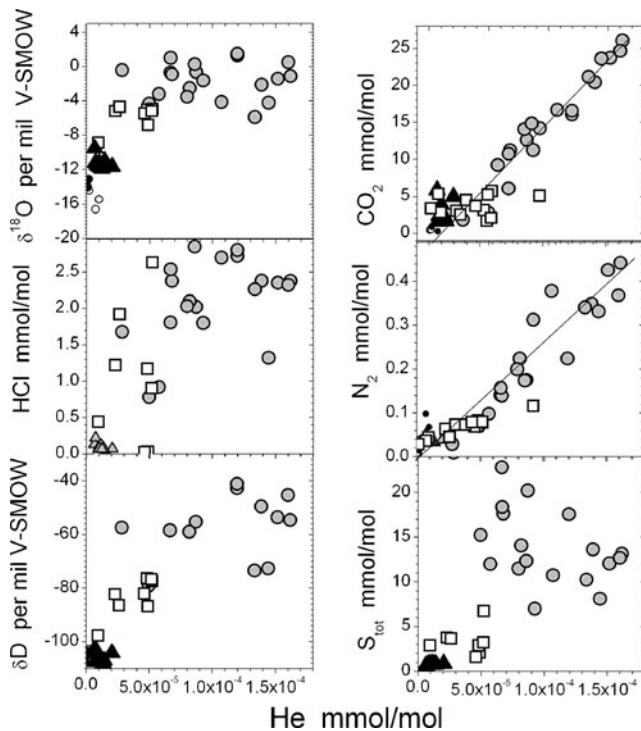


Fig. 9 Variations in water isotopes and major species in volcanic and hydrothermal gases of Mutnovsky as a function of He concentration. Lines show the apparently different mixing trends for AF and BF gases. For symbols see Fig. 3

are intermediate between almost “arc-magmatic”, hot AF gases and “hydrothermal” UF gases.

Active Funnel with ~200 t/d of the permanent SO_2 flux is the orifice of magmatic conduit with a convecting magma, similar to other open-conduit volcanoes like White Island, Satsuma Iwojima, Kudryavy, (Giggenbach 1987; Kazahaya et al. 1994; Stevenson and Blake 1998; Botcharnikov et al. 2003). Discharge rate and composition of gases from AF do not vary significantly over the last 50 years. Relatively low $^3\text{He}/^4\text{He}$ ratios in AF gases may indicate that an intermediate, crustal magma chamber is the main contributor of magmatic fluids for the volcano, and some amount of radiogenic ^4He is admixed from the continental crust. Therefore, the Bottom Field fumaroles should be connected with another magma body with a much smaller or no contribution from the crustal fluid. Another possibility is that the observed $^3\text{He}/^4\text{He}$ ratios of the AF gases may not relate to the direct crustal contamination but to be a characteristic of the deep arc-magmatic fluid of this part of the Kamchatka arc.

All workers have noted a complicated nature of the Mutnovsky plumbing system and a large variety of the erupted products (Marenina 1956; Melekestsev et al. 1987; Selyangin 1993; Duggen et al. 2007). Duggen et al. (2007), based on the Pb-isotopes and trace elements study in lavas of Mutnovsky and the nearest Gorely volcanoes, found that some Mutnovsky volcanic front magmas are “polluted”

with Gorely rear-arc melt component (with a higher fraction of the mantle material) due to specific conditions of the mantle wedge convection. It may be suggested, therefore, that a batch of this “polluted” magma is responsible for the high $^3\text{He}/^4\text{He}$ and lower N_2/Ar and N_2/He ratios in the Bottom Field fumaroles. Additionally, the Bottom Field may discharge gases from more degassed magmatic source than AF fumaroles. This may be supported by the ternary $\text{S}_{\text{tot}}\text{-CO}_2\text{-HCl}$ plot (Fig. 3b) where AF and BF gases clearly demonstrate different S/Cl ratios and a relative enrichment of BF gases in Cl. Partially degassed magma is usually enriched in Cl, which is much more soluble than CO_2 and sulfur (Carroll and Webster 1994). The enrichment in Cl is accompanied with the depletion in deuterium of the releasing from magma water vapor (Villemant and Boudon 1999; Taran et al. 2002). However, we cannot prove this in our case due to a large scattering of the water isotopic composition of the Bottom Field fumaroles and a high proportion of meteoric vapor (Figs. 4 and 12). Moreover, the lower relative S content in BF fumaroles may be related to precipitation of native sulfur taking into account the lower temperatures compared with AF gas discharges.

Hydrothermal fluids from wells and steam vents are characterized by much lower He concentrations ($\text{He}/\text{Ar} \sim 0.005$) but still high R/Ra of 4.5–5.5, typical for many other high-temperature hydrothermal systems with similar tec-

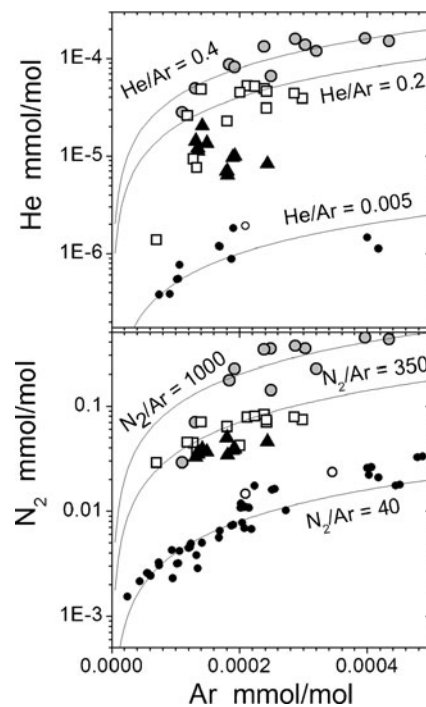


Fig. 10 Variations in N_2 and He in volcanic and hydrothermal gases of Mutnovsky as a function of their Ar content. Curves showing different He/Ar and N_2/Ar are the regression lines for the corresponding clusters of points. Symbols as in Fig. 3

tonic setting. Fluids from geothermal wells of Mutnovsky as a rule are more depleted in ^3He than steam vents.

Nitrogen Isotopes

Table 2 shows a limited set of our data on nitrogen isotopes. All data are plotted in Fig. 14 against $\text{N}_{2,\text{ex}}$ where subscript “ex” stands for the fraction of excess nitrogen, i.e. non atmospheric N_2 . The fraction of the non-atmospheric nitrogen in the samples can be estimated using $\text{N}_2/^{36}\text{Ar}$ ratio measured simultaneously with the N isotopic composition. (Inguaggiato et al. 2006). Two estimates of the fractions of non-atmospheric N_2 are shown in Table 2: one for the direct air contamination and another one for the ASW nitrogen. It can be seen that all hydrothermal gases are characterized by $\text{N}_2/^{36}\text{Ar}$ between the ASW value of 11200 and the air value of 24660. Therefore, values within this range have positive N excess relative ASW and negative one relative air (Table 2). Most of the hydrothermal samples have $\delta^{15}\text{N}$ close to zero.

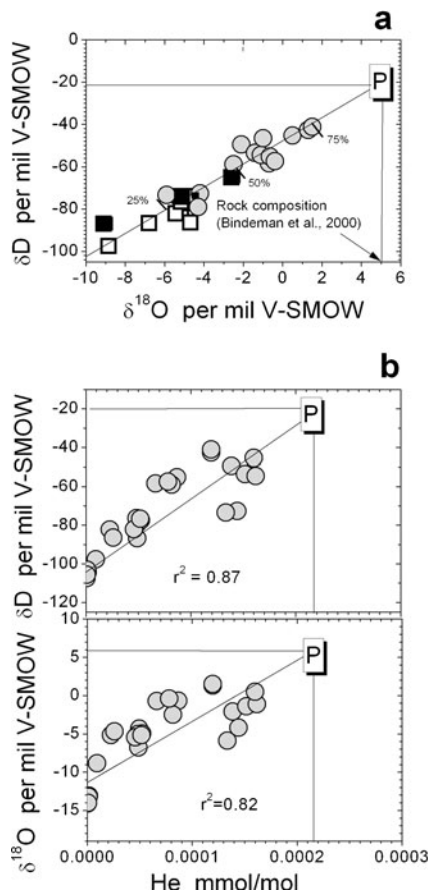


Fig. 11 a Evaluation of δD of the parent magmatic gas of Mutnovsky using extrapolation of data for Active Funnel and Bottom Field to the $\delta^{18}\text{O}$ value of +5.9‰ (average rock composition from Bindeman et al. 2004). b Evaluation of He concentration in the Mutnovsky parent gas (mmol/mol) using water isotopes. Black squares are BF data from Taran et al. (1992). Fractions of parent gas (%) are also shown on the panel (a). For other symbols see Fig. 3

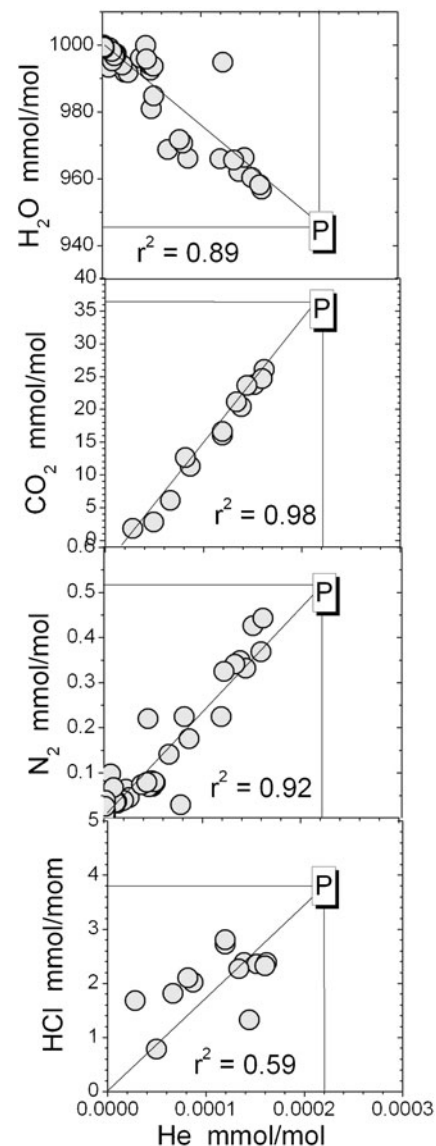


Fig. 12 Evaluation of the parent gas composition (CO_2 , HCl , H_2O and N_2) using linear regression of their correlation with He

The Upper Field and Bottom field fumaroles in spite of high N_2/Ar (and $\text{N}_2/^{36}\text{Ar}$) have $\delta^{15}\text{N}$ also close to zero or slightly negative. One of geothermal wells (045) discharges N_2 with a negative $\delta^{15}\text{N}$ of -3.4‰ . Li et al. (2009) reported a strong kinetic isotopic fractionation of N between NH_3 and N_2 with depletion of N_2 in ^{15}N relative NH_3 of $\sim 35\text{‰}$ at $600\text{--}700^\circ\text{C}$. It may be suggested that the NH_3/N_2 ratio is an important factor controlling $\delta^{15}\text{N}$ of N_2 . Indeed, the NH_3/N_2 ratios shown in Table 2 are on average very low for AF, but close to 1 in the BF, UF and hydrothermal gases. The Active Funnel gases with the highest $\text{N}_2/^{36}\text{Ar}$ (and N_2/Ar) values have positive $\delta^{15}\text{N}$ with a tendency to have higher $\delta^{15}\text{N}$ in samples with a higher fraction of the non-atmospheric nitrogen. This is typical for many arc-type volcanoes and is considered to be a proof for the volatile contribution from the subducted

Table 4 Composition of parent gas (AF*) calculated for gases from the Active Funnel (see text for details) and mean compositions of gases from other sites. Concentrations in mmol/mol

	AF*	<BF>	<UF>	<SV>	<W>
H ₂ O	943	994±3	996±1.5	998.8±0.99	999.4
CO ₂	36	3.72±1.32	2.89±1.42	1.16±0.9	0.53±0.27
HCl	3.2				
S _{tot}	17.2				
N ₂	0.52	0.067±0.022	0.038±0.05	0.0595±0.035	0.016±0.014
Ar	5×10 ⁻⁴	(1.8±0.8)×10 ⁻⁴	(1.7±0.4)×10 ⁻⁴	0.0011±0.0007	(2.6±2.2)×10 ⁻⁴
He	2.2×10 ⁻⁴	(3.8±2.3)×10 ⁻⁵	(1.1±0.4)×10 ⁻⁵	(5±3)×10 ⁻⁶	(1.1±0.5)×10 ⁻⁶
CH ₄	3×10 ⁻⁴	0.0023±0.0007	0.022±0.003	0.011 ±0.006	0.0012±0.001
δD ‰	-19	-80.4±8.4	-106±1.8	-105 ±2	-105±2
δ18O ‰	+5.9	-5.6±1.8	-11.3±0.7	-15.5 ±1	-13.6±0.4
N ₂ /Ar	1040	372	224	54	62
N ₂ /He	2364	1763	3455	11900	14545
³ He/ ⁴ He (R/Ra)	6.6	7.67	6.3	5.7	5.4
He/Ar	0.44	0.211	0.065	0.005	0.004
N ₂ /CH ₄	1733	29.1	1.7	5.4	13.3
CO ₂ /N ₂	62	56	76	19	33
CO ₂ /He	163636	97895	262727	232000	481818
CO ₂ / ³ He	1.8×10 ¹⁰	0.9×10 ¹⁰	3×10 ¹⁰	2.9×10 ¹⁰	6.4×10 ¹⁰
N ₂ / ³ He	2.6×10 ⁸	1.6×10 ⁸	3.9×10 ⁸	1.5×10 ⁹	1.9×10 ⁹
Cl/ ³ He	1.6×10 ⁹	2.6×10 ⁹	0.9×10 ⁹	–	–

sediments (Sano et al. 2001; Fischer et al. 1998, 2002; Hilton et al. 2002; Inguaggiato et al. 2006; Inguaggiato et al. 2009).

Sources of the Mutnovsky gases

For arc volcanoes, several sources of fluid can be specified (e.g. Hilton et al. 2002; Wallace 2005; Taran 2009 and references therein). As the first approximation they are: upper mantle (MORB), subducted slab (S) and air saturated meteoric water (ASW). We do not take into account here air contamination during sampling or analysis. Main sources of volatiles from the subducted slab are oceanic sediments (SED) and a layer of altered oceanic basalts (AOC). An important source of volatiles may be the continental crust beneath the volcano edifice (CC). Altogether there are five endmembers (MORB-AOC-SED-CC-ASW) or four if an integrated slab endmember is considered (S = SED + AOC). We use the term “MORB” for the composition of the mantle in the mantle wedge and assume that this sub-arc asthenospheric mantle has MORB characteristics in terms of volatile composition.

Table 5 shows mean elemental ratios and isotopic compositions of CO₂, N₂, Ar and He in main reservoirs, potential contributors of volatiles to volcanic gases of subduction zones. Characteristics of crustal gases (old metamorphic crust) are taken from KTB deep well in Germany (Bach et al. 1999; Lippmann et al. 2005). The only data on He and Ar for altered oceanic crust were

reported by Staudacher and Allegre (1988). They have also reported the abundances of Ar and He in a few samples of pelagic oceanic sediments, isotopic compositions of Ar and He in altered basalts and only Ar isotopes in sediments. The published data on He abundances and isotopes in pelagic sediments (e.g. Ozima and Podosek 2002, for review) show very low He content and very high ³He/⁴He (30–100Ra) due to a constant (though at a very low rate) contribution from the cosmic dust (IDP—interplanetary dust particles). Hilton et al. (2002) have discussed the importance of these data for He balance in subduction zones and decided do not count this excess of ³He in sediments referring to a probable loss of He from sediments before subduction by diffusion. Nitrogen in oceanic sediments and oceanic crust was analyzed by Sadofsky and Bebout (2004), Li and Bebout (2005), Li et al. (2007) and Busigny et al. (2005). The “Slab” component in Table 5 is presented as a mean composition with high N₂/Ar, N₂/He and a low ³He/⁴He. A reason for this “averaging” is that it is impossible to distinguish contribution to N₂-Ar-He separately from sediments and AOC due to large uncertainties in Ar and He concentrations.

Two plots (modified from Taran 2009) with mixing lines between main potential reservoirs of N₂, Ar and He are shown in Fig. 15. The point for the evaluated parent gas (AF) is very close to the mixing lines between slab and mantle. Other points for volcanic gases (BF and UF) are shifted to the mantle-atmosphere mixing lines. Hydrothermal gases occupy positions closer to the mantle-atmosphere mixing lines.

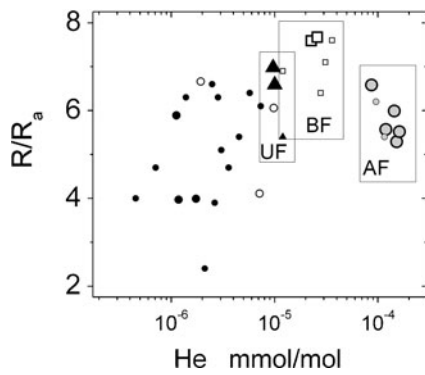


Fig. 13 Helium isotopic composition as a function of He concentration in gases of Mutnovsky. Small symbols are from Taran et al. (1986) and Rozhkov and Verkhovsky (1990)

Distinguishing between various sources of volatiles (mantle wedge, subducted slab, sub-arc crust) is possible by considering the four coupled components (CO₂, N₂, Ar, He) altogether. In that case the concentration of each component of the parent magmatic gas X_i is a weighed sum

$$X_i = \sum f_j X_{ij}, \tag{1}$$

where f_j is the fraction of fluid from each source (endmember) and X_{ij} is the concentration of each specie normalized by the sum CO₂ + N₂ + Ar + He. If there are four sources (reservoirs): M—mantle; S—slab; C—crust, and A—atmosphere (ASW), then we have four equations for four variables. Taking into account the mass balance equation f_M + f_S + f_C + f_A = 1, we can use three equations for concentrations and one for the mass balance. This approach is considered in details by Albarede (1995). If all parameters are chosen correctly, the solution (f_j) of the system with the observed or presumed composition of the parent (or any) gas X_i is a set of f_j values with 0 ≤ f_j ≤ 1. This solution should be the same for any set of three equations for concentrations, for example, (CO₂, He, Ar), (CO₂, N₂, He), (N₂, He, Ar), etc. In practice, all parameters (concentrations) of endmembers are mean values with large standard deviations. Moreover, the slab endmember in terms of CO₂ + N₂ + Ar + He is poorly determined because of the unknown fraction of carbon subducted to the lower mantle (Kerrick and Connolly 2000; Poli and Schmidt 2002). The observed or presumed gas compositions at the surface have also certain errors or ranges of values. Therefore, this “ideal” approach for the system (CO₂-N₂-Ar-He) does not work, giving either different reasonable solutions for different sets of equations or solutions that do not make sense (f_j negative and/or >1). Nevertheless, the set of endmember compositions presented in Table 5 for the N₂-Ar-⁴He-³He mixture (CO₂ is replaced for ³He) is “compatible” with any composition of Mutnovsky gases shown in Table 4. This means that any combination of equations: N₂-Ar-⁴He, N₂-⁴He-³He, Ar-⁴He-³He, N₂-Ar-³He-⁴He—derives almost identical solutions (f_j) for a given composition of the

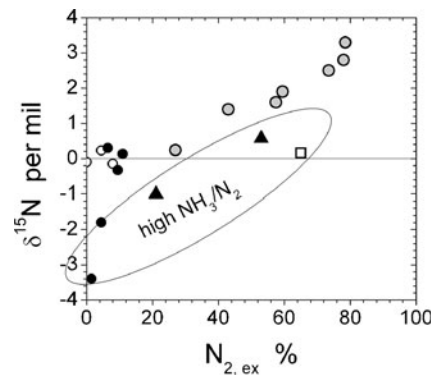


Fig. 14 Isotopic composition of nitrogen as a function of excess N₂ calculated from the N₂/³⁶Ar ratio relative ASW (Table 2). See text for explanation. Symbols as in Fig. 3

gas. Table 6 shows fractional contributions to the N₂-Ar-He system of Mutnovsky gases from mantle, slab, continental crust and air-saturated water calculated by using this method. Only ~1% (mole basis) of the N₂+ Ar + He mixture enter to the parent Mutnovsky gas from the mantle wedge. But this 1% of N₂ + He + Ar carrying almost 1% of the mantle N₂, contains more than 80% of the total mantle He and more than 95% of the total ³He. The predominant contribution of N₂ (>95%) occurred from the slab. The contribution from the crust beneath the volcano is negligible. However, even the “corrected” parent gas composition has a fraction (~3%) of ASW. The ASW fraction increases from parent gas to hydrothermal gases from steam vents and wells where it reaches more than 60%.

A system including CO₂ and carbon isotopic composition is needed to distinguish contributions from subducted sediments and oceanic crust. In a three end-member model consisting of MORB (M), slab-derived marine carbonate (L) and organic carbon (S) used by a number of authors (Hilton et al. 2002; Shaw et al. 2003 and references therein), carbon isotopic composition of CO₂ and CO₂/³He are the most popular parameters for estimations of the fraction contributed from each reservoir. The following mass balance equations were first used by Sano and Marty (1995):

$$\delta^{13}C_o = f_M \delta^{13}C_M + f_L \delta^{13}C_L + f_S \delta^{13}C_S \tag{2}$$

$$\left(\frac{{}^3\text{He}}{\text{CO}_2}\right)_o = f_M \left(\frac{{}^3\text{He}}{\text{CO}_2}\right)_M + f_L \left(\frac{{}^3\text{He}}{\text{CO}_2}\right)_L + f_S \left(\frac{{}^3\text{He}}{\text{CO}_2}\right)_S \tag{3}$$

$$f_M + f_L + f_S = 1 \tag{4}$$

This approach is fruitful for the Kamchatkan gases because subducting oceanic sediments close to the Kamchatka trench do not contain carbonates (Plank and Langmuir 1998; Hilton et al. 2002; Jarrard 2003). Therefore, it can be suggested that all carbonate carbon contributing to magmatic gases of

Table 5 Relative concentrations and isotopic compositions of N₂, Ar and He in different terrestrial reservoirs, potential contributors to the subduction-type magmas, in parent gas of Mutnovsky and in some subduction-related volcanic gases. Vulcano M corresponds to mag-

matic component in gases as defined by Taran (2011). The uncertainties (relative errors) are given very approximately, just to give an idea about the natural scatter of data. Error of 100% indicates that the observed variations may be higher than one order of magnitude

Reservoir	N ₂ /Ar	N ₂ /He	He/Ar	R/Ra	N ₂ / ³ He	CO ₂ /N ₂	CO ₂ /He	Error %	Ref
MORB	100	25	4	8	2.3E+08	1000	2.2E+04	30	1–3
AOC	10000	20000	0.5	0.1	1.4E+11	60	1E+06	100	3,4
SED	40000	2E+05	0.2	??	2.2E+11	60	1E+07	100	3,4
KTB	700	100	7	0.07	4.2E+09	0.00033	0.035	30	5,6
Mutnovsky P	1040	2364	0.44	6.59	1.9E+08	69	1.8E+05	20	7
Kudriavy P	380	2100	0.18	6.70	2.2E+08	72	1.5E+05	50	8
Vulcano M	470	470	1	6.2	5.4E+07	140	6.7E+04	50	9
Air	83.6	1.49E+05	5.61E-04	1	1.1E+11	0.00045	67		
ASW 5°C	38	2.64E+05	1.46E-04	1	1.9E+10	0.015	1000		

References: 1—Marty and Humbert (1997); 2—Marty and Zimmermann (1999); 3—Hilton et al. (2002); 4—Staudacher and Allegre (1988); 5—Bach et al. (1999); 6—Lippmann et al. (2005); 7—this work; 8—Fischer et al. (1998); 9—Taran (2011)

Mutnovsky is derived from the oceanic crust (AOC), and all organic fraction is derived from oceanic sediments. The only data on $\delta^{13}\text{C}$ of CO₂ in Mutnovsky gases were published by Taran (1988) for the hydrothermal system and Taran et al. (1992) for crater fumaroles. A mean value for $\delta^{13}\text{C}$ -CO₂ of all Mutnovsky gases may be estimated as $-9\pm 2\%$ (V-PDB), i.e. they are relatively enriched in ¹²C compared with the mantle values in the -5% to -7% range (e.g. Javoy et al. 1986). Relatively isotopically light CO₂ is common for Kamchatkan volcanoes (Taran 1992) and for volcanoes from some other subduction zones like Cascadian and New Zealand (Taylor 1986).

We accept CO₂/³He ratios of 2×10^9 , 1×10^{13} and 1×10^{13} for M, L and S, respectively, from Hilton et al. (2002). For $\delta^{13}\text{C}$ of organic carbon a value of -25% is accepted. Solution of the system (2–4) for the parent Mutnovsky gas with $\delta^{13}\text{C}$ -CO₂ = -9% and CO₂/³He = 2×10^{-10} (Table 3) gives M=0.1; S=0.4 and L=0.5. In this case the mantle

contribution of CO₂ is ~ 10 times higher than that of N₂ compared with the N₂-Ar-He system. This result is in a qualitative agreement with conclusions of Taran (2009) who estimated the mantle contribution of CO₂ to Kamchatkan magmatic fluids as 18% and N₂ as $\sim 3\%$ (mole basis). Fischer et al. (1998) estimated 12% of the mantle CO₂ and 2% of the mantle N₂ for Kudryavy gases which is very close to our values for Mutnovsky. Application of the method to BF gases gives a much higher mantle contribution for CO₂ (32%) and same L/S (or AOC/Sed) ratio. This result together with a significantly higher ³He/⁴He in BF fluids may indicate the presence of a separate degassing magma body for the Bottom Field fumaroles. Application of the method to UF volcanic gases and hydrothermal gases does not make much sense because UF gases are strongly affected by hydrothermal fluids, and CO₂ in hydrothermal fluids is controlled by thermodynamic conditions of the hydrothermal aquifer (Giggenbach 1980, 1997b; Taran 1986).

Fig. 15 Plots N₂/He vs N₂/Ar (a) and R/Ra vs He/Ar (b) for volcanic and hydrothermal gases of Mutnovsky. Shaded areas correspond to compositional variations of end-members. Point AF is plotted as for the parent gas with the extracted meteoric component (Table 4). Points UF and BF are mean values for the compositions of gases from UF and BF (Table 4)

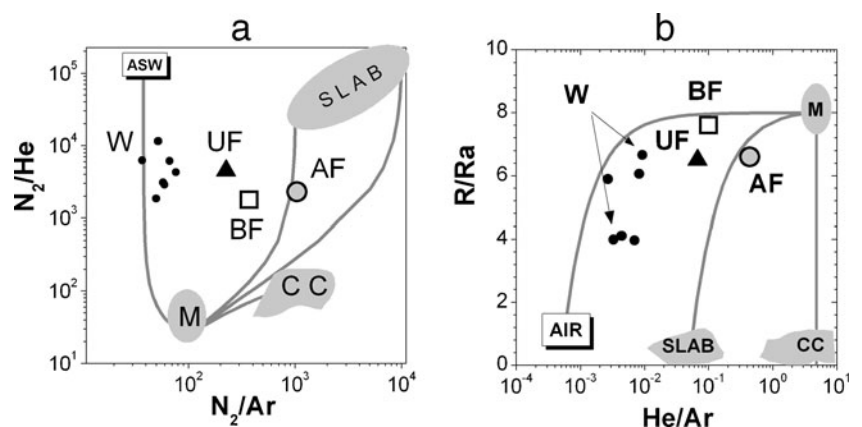


Table 6 Fractional contributions of MORB (M), slab (S), continental crust (C) and atmosphere (A) to N₂, Ar and He in volcanic and hydrothermal gases of Mutnovsky. The fractions are obtained as a solution of equations (1) with different sets of species. AF* parent gas

	AF*	<BF>	<UF>	<SV>	<W>
N ₂ +He+Ar					
M	0.01	0.014	0.007	0.0001	0.001
S	0.96	0.89	0.83	0.297	0.39
C	0.00013	0.0025	0.0003	0.014	0
A	0.031	0.095	0.17	0.69	0.61
N ₂ + ³ He+ ⁴ He					
M	0.0091	0.014	0.0059	0.0015	0.001
S	0.96	0.89	0.82	0.306	0.39
C	0.0052	0.00014	0.0042	0.0017	0.0013
A	0.031	0.095	0.17	0.69	0.61
N ₂ +Ar+ ⁴ He+ ³ He					
M	0.0091	0.014	0.0059	0.0015	0.0011
S	0.90	0.88	0.81	0.25	0.34
C	0.0053	0	0.004	0.0018	0.0014
A	0.085	0.10	0.18	0.75	0.66
Ar+ ⁴ He+ ³ He					
M	0.0091	0.014	0.0059	0.0015	0.0011
S	0.90	0.73	0.54	0.25	0.34
C	0.0053	0	0.005	0.0018	0.0015
A	0.085	0.26	0.45	0.75	0.66

Concluding Remarks

- (1) To provide the existing compositional and thermal distribution of fumarole discharges at Mutnovsky, there must be at least three independent deep sources of fluid: (1) a large source of magma with a continuous high flux of volatiles (80% of the total flux) through the open conduit of the Active Funnel; (2) a distinct magma body with higher ³He/⁴He, lower CO₂/³He and N₂/³He feeding the Bottom Field; (3) a high-temperature hydrothermal fluid with a high CH₄ content, which is the predominant source for the Upper Field gases. These sources of fluids are schematically shown in Fig. 16 on the hypothetical North-South cross-section of the volcanic edifice and the adjacent hydrothermal aquifer of Mutnovsky.
- (2) The Active Funnel emits the hottest and most concentrated gas that requires a hot convecting magma body. Its parent magmatic gas characteristics are typical of a subduction-related magmatic gas, with the main volatile contribution from the subducted slab. This magma body must be of significant size. Taran et al. (1992) have estimated the volume of the completely degassed melt necessary to ensure the observed emission rate at ~4 km³ per 100 years. It is not clear whether this

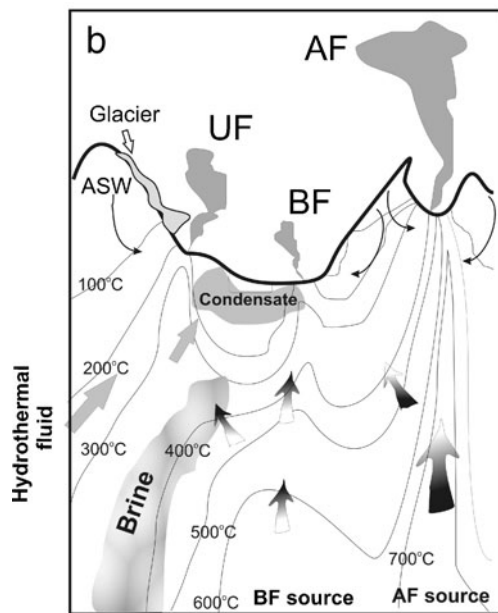
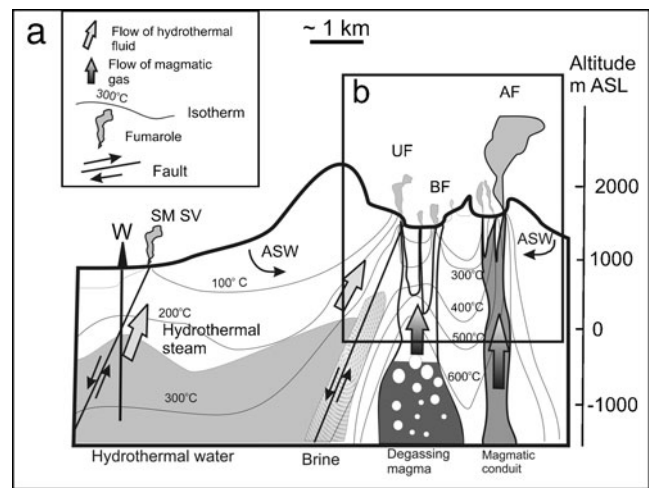


Fig. 16 **a** A conceptual model of the volcano-hydrothermal system of Mutnovsky. A cross-section is drawn through Mutnovsky crater (appx SW-NE) and the Mutnovsky edifice (appx N-S) showing temperature distribution and main fluid flows. **b** Fluid flows and temperature distribution beneath the crater. AF, BF, UF are Active Funnel, Bottom and Upper Fields, respectively. SM-SV denotes “Severo-Mutnovsky steam vents”, W—geothermal well

- magma chamber is related to a large intrusive body whose presence beneath the volcano and hydrothermal system has been deduced from results of the geophysical survey (Selyangin and Ponomareva 1999).
- (3) The Bottom Field is probably fed from a separate magma body whose volatile characteristics are more similar to those of rear-arc magmas with a larger contribution of volatiles from the mantle wedge. It is likely to be a shallow intrusive body of crystallizing basaltic magma responsible for the last basaltic eruptions of Mutnovsky and “contaminated” by the

rear-arc component as suggested by Duggen et al. (2007). Fumarolic gases of the Bottom Field are influenced by shallow hydrothermal processes because of the existence of a liquid phase (brine?) beneath the crater (Fig. 16b). This liquid phase is most probably a partially evaporated mixture of the volcanic gas condensate with meteoric water and the condensate of hydrothermal steam feeding the Upper field fumaroles. Subsurface boiling of this highly acidic condensate may provide a wide range of HCl and S concentrations in volcanic vapors and strong variations in the vapor isotopic compositions.

- (4) The mixed fluid venting through the Upper Field is most probably controlled by a fault, contains only <2% of the parent magmatic gas and is composed of hydrothermal fluid. A notable feature of the Upper Field hydrothermal gas is an elevated CH₄ content which is higher than in the fluid of the main geothermal field. The high-CH₄ volcanic-hydrothermal fluid discharging from the Upper Field fumaroles may partially represent a secondary component, a vapor derived from a deep, two-phase, saline brine-vapor “envelope” formed near the contact between hydrothermal aquifer and the magmatic system of Mutnovsky, similar to what has been suggested by Giggenbach (1987) for White Island.
- (5) The nearest hydrothermal manifestations northward of Mutnovsky are separated by a distance of 5 km from the crater fumarole fields (Fig. 2). They are very similar to the UF gases in terms of the isotopic composition of water vapor, with the difference in the ¹⁸O shift and a lower CH₄ concentration. Three main features of the UF gases are not simple to explain: (1) their elevated methane content which is higher than in hydrothermal fluid from steam vents and wells; (2) a high fraction of non-atmospheric nitrogen (N₂/Ar~230); (3) the high temperature of fumaroles, up to 280°C, that requires an additional heat source to maintain the temperature of the deep hydrothermal fluid to the surface. The heat problem might be resolved assuming that the heat source can be the same shallow basaltic intrusion beneath the main crater, which is the source of BF gases (Fig. 16). We speculate that the excess CH₄ and non-atmospheric N₂ are derived from a non-magmatic source at the interface between the magmatic and hydrothermal system (hydrothermal “envelope”), most probably from the deeper part of the volcano-hydrothermal system. On the other hand, ~2% of the parent gas or ~12% of the BF gas added to hydrothermal fluid represented by SV gases can provide the required N₂/Ar ratio for UF gases. In this case the non-atmospheric N₂ and CH₄ have different sources (magmatic and crustal).

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